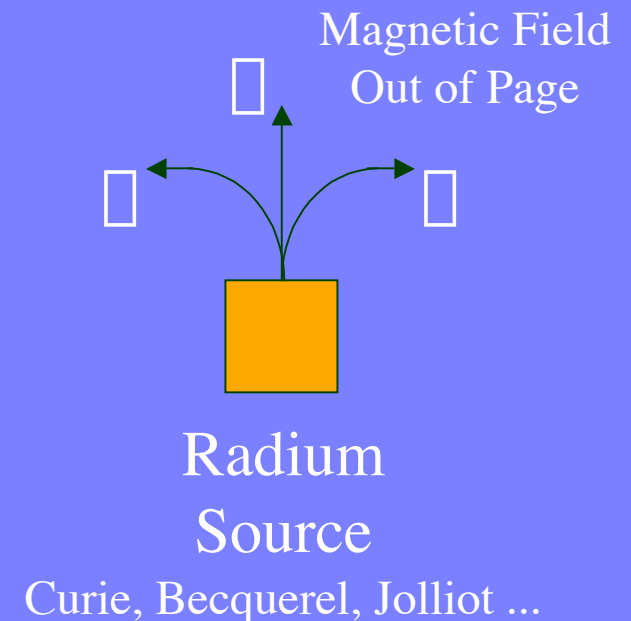


# Lecture 7: Radiation and Health

- Types of ionizing radiation

- Alpha particles  ${}^4_2\text{He}^{++}$  “nuclei”
- Beta particles negative electrons
- Gamma particles: photons
  - X-ray energies 10's-100's keV
  - $\gamma$ -ray energies: 500 keV and higher
- Neutrons ( ${}^1_0\text{n}$  “nuclei”) (discovered by Chadwick - 1932)



# Alpha Particles

Produced in decay of **HEAVY** nuclei:  $^{226}\text{Ra} \rightarrow ^{222}\text{Rn} + ^4\text{He}$

Conservation of CHARGE (Z)

$$88 = 86 + 2$$

Conservation of Nucleons (A)

$$226 = 222 + 4$$

Conservation of ENERGY

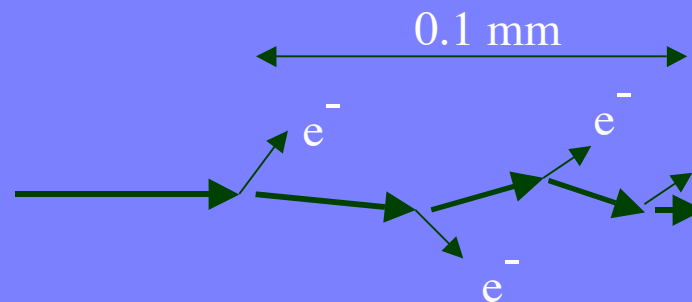
$$m(^{226}\text{Ra})c^2 = m(^{222}\text{Rn})c^2 + m(^4\text{He})c^2 + \square$$

Nuclear Energy  
(MeV's)

Other alpha decays:  $^{238}\text{U} \rightarrow ^{234}\text{Th} + \square$  ( $t_{1/2} = 4.5 \times 10^9$  y)

$^{232}\text{Th} \rightarrow ^{228}\text{Ra} + \square$  ( $t_{1/2} = 1.4 \times 10^{10}$  y)

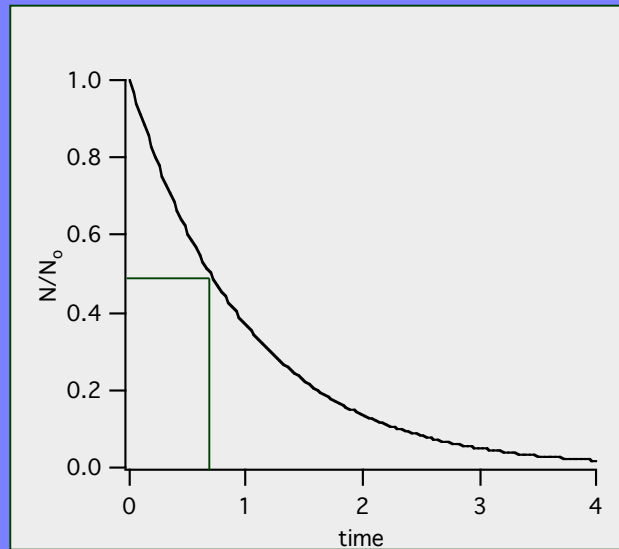
Alphas lose energy by IONIZATION



Short Range  
High Specific Ionization/  
Energy Deposited

# Half life

Radioactive decay:  $\frac{dN}{dt} = -N/\tau$      $N=N_0 e^{-t/\tau}$  (t is time)



$$t_{1/2} = \tau \ln 2 = 0.69\tau$$

# Beta Decays

Electrons and Positrons are Beta-particles

$$m = 9.1 \times 10^{-31} \text{ kg}; mc^2 = 0.511 \text{ MeV}$$

$$q = \pm 1.6 \times 10^{-19} \text{ Coulombs}$$

Neutron Rich Isotopes  $\beta^-$  decay:  $^{234}\text{Th} \rightarrow ^{234}\text{Pa} + e^- + \bar{\nu}$  ( $t_{1/2} = 24.1 \text{ d}$ )

$$\text{Conservation of CHARGE (Z)} \quad 90 = 91 - 1$$

$$\text{Conservation of Nucleons (A)} \quad 234 = 234 + 0$$

$$\text{Conservation of ENERGY} \quad m(^{234}\text{Th})c^2 = m(^{234}\text{Pa})c^2 + m(e^-)c^2 + \bar{\nu}$$

Betas are PENETRATING (similar to gamma rays)

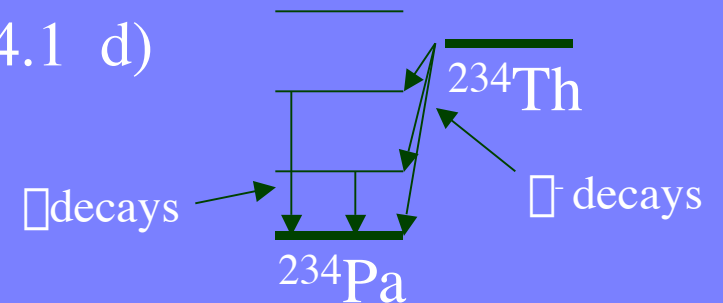
Longer Range than Alpha Particles

Lower Specific Energy Loss

# Gamma Rays (Photons)

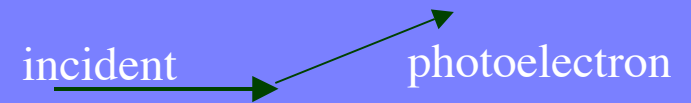
Produced by Brehmstrahlung (of high energy electrons)

Produced in NUCLEAR TRANSITIONS

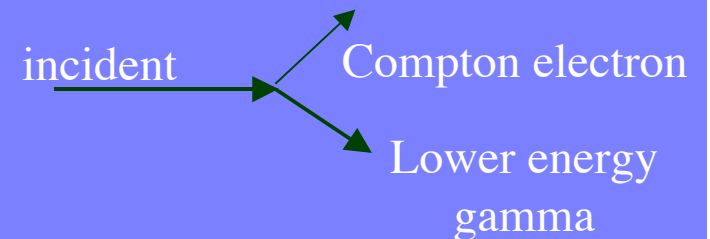


Photon interactions:

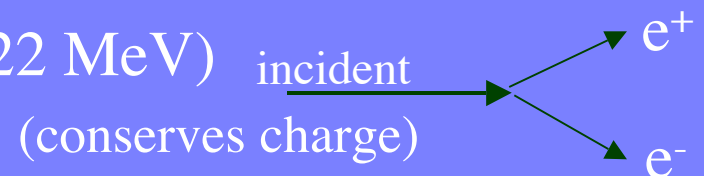
Photoelectric effect (ionizing)



Compton Scattering (ionizing)



Pair Production (for  $E_\gamma > 1.022 \text{ MeV}$ )



(conserves charge)

# Penetration and Shielding

Ionizing radiation interacts with materials

Alphas lose energy by ionization

Range (to KE=0)  $\sim m/Z \ll 1$  mm

Betas lose energy by ionization

range  $\sim 1/Z$

Gammas (photons) transfer energy to electrons

Attenuation ( $e^{-\mu x}$ ) depends on  $Z$ ,  $E_\gamma$

# Harmful Effects of Ionizing Radiation

- Ionizing Radiation: dislocates electrons/atoms
  - $\text{H}_2\text{O} + \text{radiation} \rightarrow \text{H}_2\text{O}^+ + \text{e}^- : \text{OH}, \text{H}^+, \text{H}, \text{OH}^-, \text{H}_2\text{O}_2$ 
    - Early cell death
    - Inhibited cell division
    - Genetic modification of chromosomes
- Units of exposure:
  - Becquerel/ Curie (activity)  $1 \text{ Bq} = 1/\text{s}$        $1 \text{ Cu} = 3.7 \times 10^{11} \text{ Bq}$
  - Gray/Rad (absorbed dose)  $1 \text{ Gr} = 1 \text{ J/kg}$      $1 \text{ rad} = 0.01 \text{ Gr}$
  - SV/Rem (includes biological effectiveness - RBE)
  - RBE's
    - 1 for x-rays
    - 5 slow neutrons
    - 20 for alpha particles, fast neutrons, protons

# Dose Calculations

Alphas: 1  $\mu$ Curie of 5 MeV alpha emitter (ingested)  
(biological half-life =  $t_{1/2} = 1$  day)

$$5 \text{ MeV} = 5 \times 10^6 \times 1.6 \times 10^{-19} \text{ J} = 8 \times 10^{-13} \text{ J}$$

$$1 \mu\text{Cu} = 3.7 \times 10^4 \text{ Gr} = 1.33 \times 10^7 / \text{hour} = 3.2 \times 10^8 / \text{day}$$

$$\begin{aligned} \text{Whole Body (70 kg) Dose} &= (8 \times 10^{-13} \text{ J})(3.2 \times 10^8 / \text{day})(1.44 t_{1/2}) / (70 \text{ kg}) \\ &= 5.3 \mu\text{Gr} = 0.5 \text{ mRad} \end{aligned}$$

RBE for alphas = 20 so EFFECTIVE DOSE = 10 mRem

**DANGER:** energy can be deposited in more **localized** region



# Allowed Dose and General Exposure

Allowed Dose: 5 mSv = 500 mRem per Year (General Public)

Natural Radioactivity	Total	299 mRem/Y
Local Gamma Rocks etc.		29
$^{14}\text{C}$ ( $t_{1/2} = 5739$ y)		1
Airborne radon		200
$^{40}\text{K}$ (0.117%/t <sub>1/2</sub> = 1.3x10 <sup>9</sup> y)		20 (bannanas, etc.)
U, Th in body		19
Cosmic Rays		30
doubles each 6000 ft: @ 36,000 feet 64x about 1mR/kmile		
Manmade Radioactivity		
Diagnostic X-rays		40
Radioactive waste		0.2
Fallout (atmospheric testing)		0.1



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Cross Word

## Calculate Your Radiation Dose

### Directions

1. Enter values or select entries where options are provided. (Some entries for the annual dose calculator are already filled in.)
2. Click the "calculate dose" button.

Source	Your Average Annual Dose (mrem)
Cosmic radiation at sea level (from outer space)	26
What is the elevation (in feet) of your town?	up to 1000 (2 mrem)
<b>Terrestrial (from the ground):</b>	
What region of the US do you live in?	Elsewhere in the US (46 mrem)
<b>Internal radiation (In your body):</b>	
From food and water, (e.g., potassium)	40
From air, (radon)	200
Do you wear a plutonium powered pacemaker?	No (0 mrem)

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Do you have porcelain crowns or false teeth? Yes (.07 mrem)

### Travel Related Sources:

Add 1 for each 1000 miles traveled by jet this year: 25

Are X-ray luggage inspection machines used at your airport? Yes (.002 mrem)

Do you use gas lantern mantles when camping? No (0 mrem)

### Miscellaneous Sources:

Weapons test fallout 1

House construction: Do you live in a stone, brick, or concrete building? No (0 mrem)

Do you wear a luminous wristwatch (LCD)? No (0 mrem)

Do you watch TV? No (0 mrem)

Do you use a computer terminal? Yes (.1 mrem)

Do you have a smoke detector in your home? Yes (.008 mrem)

How many medical x-rays do you receive per year? (40 mrem each) 1

medical  
procedures do you  
receive per year?  
(14 mrem each)

Do you live within  
50 miles of a  
nuclear power  
plant?

Do you live within  
50 miles of a coal  
fired power plant?

TOTAL YEARLY DOSE (in mrem)

#### Notes

- The amount of radiation exposure is usually expressed in a unit called millirem (mrem). In the United States, the average person is exposed to an effective dose equivalent of approximately 360 mrem (whole-body exposure) per year from all sources (NCRP Report No. 93).
- The dose calculator is based on the American Nuclear Society's brochure, "Personal Radiation Dose Chart". The primary sources of information we relied on are the National Council on Radiation Protection and Measurements Reports #92-#95, and #100. Please remember that the values used in the calculator are general averages and do not provide precise individual dose calculations.

 National Council on Radiation  
Protection and Measurements  EPA

- Special thanks to UT-ANS group for providing the CGI script that was used to construct this yearly dose calculator web page.

 Radiation: Risks and Realities  
information on radiation, radiation uses and  
risks

[Understanding Radiation](#)

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# Cancer Risk

Cases per 1000 Sv ( $10^5$  Rem)

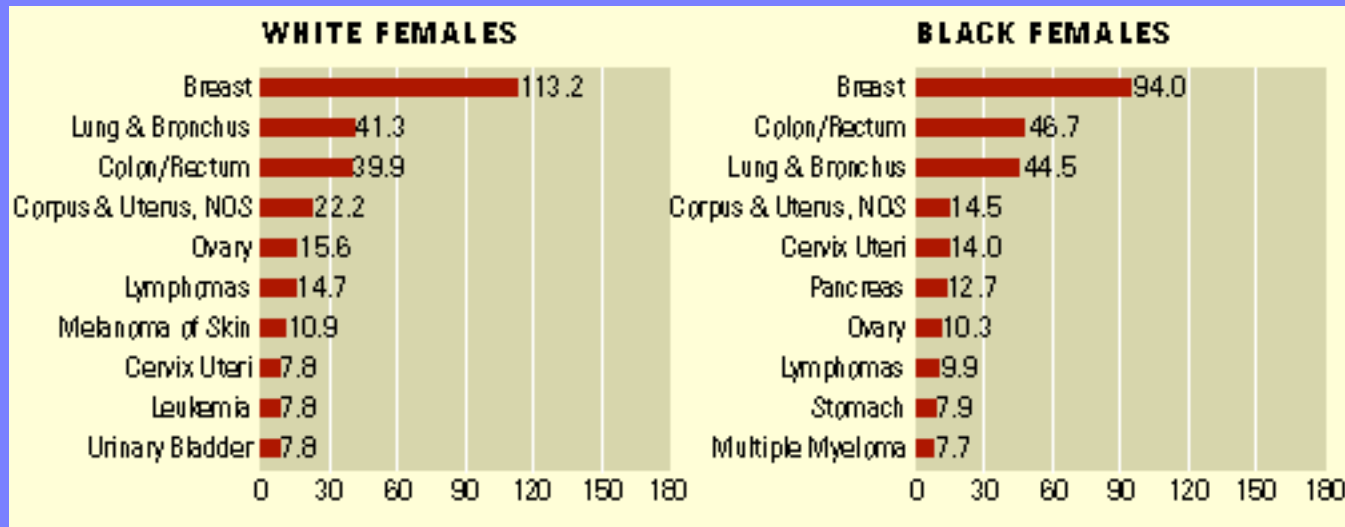
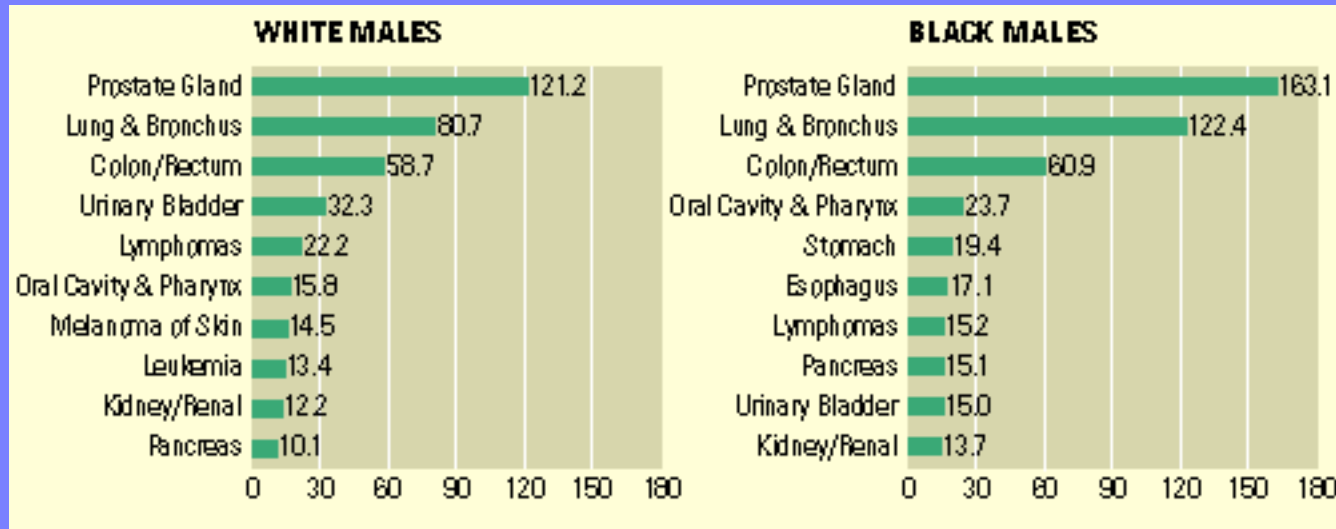
	Male	Female
Leukemia	2	2
Breast	-	5
Lung	2	2
Thyroid	0.5	0.5
Bone Sarcoma	0.5	0.5
Other	5	5
Total	10	15

Assumes the Linear model:

1000 people @ 1 SV ~ 1000000 people @ 0.001 Sv

<http://seer.cancer.gov>

# Total Cancer Rates per 100,000



Source: NIH ([http://rex.nci.nih.gov/NCI\\_Pub\\_Interface/raterisk/rates12.html](http://rex.nci.nih.gov/NCI_Pub_Interface/raterisk/rates12.html))

# Typical Dose

- Chest Xray 20 mR
- Brain Xray 250 mR
- Abdomen 550 mR
- Dental 10 mR
- Mammography 50 mR
- CT slice 1000 mR
- Liver tumor 6000 R !

# Radiation Safety

- Time:  $N = N_0 e^{-t/\tau}$
- Distance:  $I \sim 1/r^2$
- Shielding:  $I = I_0 e^{-\mu x}$
- Monitoring (radiation detection)